High Pressure Superconducting Transition in Dihydride BiH₂ with Bismuth Open-Channel Framework

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Metal hydrides MH_x ($x \le 2$) with low hydrogen content are not expected to show high- T_c superconductivity owing to the low hydrogen-derived electronic density of states at Fermi level and the limited hydrogen contribution to electron-phonon coupling strength. In this Letter, we report on the successful synthesis of a novel bismuth dihydride superconductor, Cmcm-BiH₂, at approximately 150 GPa, and the discovery of superconductivity with T_c about 62 K at 163 GPa, marking the first instance of superconductor among the MH2-type metal dihydrides. Cmcm-BiH2 adopts a unique host-guest type structure, in which the Bi atoms via weak Bi-Bi covalent bonds form a three-dimensional open-channel framework that encapsulates H₂-like molecules as guests, thereby broadening the structural diversity of hydrides under high pressures. The occurrence of superconductivity is evidenced by a sharp drop of resistivity to zero and the characteristic downward shift of T_c under applied magnetic fields. Notably, Cmcm-BiH₂ remains stable down to at least 97 GPa during decompression, with the calculated lowest pressure for dynamic stability of 10 GPa. In-depth analysis reveals that the covalent bismuth open-channel structure forms metallic conduction channels, dominates the electronic states near the Fermi level, and contributes approximately 51% of the total λ in *Cmcm*-BiH₂, distinguishing it from known high-pressure hydride superconductors. These findings highlight the critical role of nonhydrogen elements in producing superconductivity and open new avenues for the design and optimization of high- T_c hydride superconductors.

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The pursuit of high-temperature superconductors has been a central goal in condensed matter physics since the discovery of superconductivity in solid mercury in 1911 [1]. According to the Eliashberg theory [2] and Allen-Dynes equation [3] for conventional Bardeen-Cooper-Schrieffer (BCS) superconductors, the superconducting critical temperature (T_c) is determined mainly by the electron-phonon coupling (EPC) strength and the frequency of the phonons involved in EPC. These factors are quantified by the EPC constant (λ) and the average frequency of the phonon spectrum weighted by the coupling strength (ω_{log}). As a result, light-element dominant compounds, especially

hydrides, have long been considered as promising candidates for achieving high $T_{\rm c}$ due to inherently high phonon frequencies, which can not only elevate the Debye frequency and enhance the λ but also boost intrinsic $\omega_{\rm log}$ values. Recent theory-driven experimental discoveries have indeed demonstrated near-room-temperature superconductivity in hydrides under high pressures, such as SH₃ [4], LaH₁₀ [5,6], and CaH₆ [7,8]. These hydride superconductors are featured by high hydrogen contents and a three-dimensional hydrogen network, which facilities strong EPC, e.g., $\lambda \sim 2$ for LaH₁₀, via high-frequency H-H vibrations [9,10]. Thus, the synergic enhancement of λ and $\omega_{\rm log}$ based on the hydrogen sublattices is essential for the achievement of high $T_{\rm c}$ in these superhydrides superconductors.

Motivated by this concept, we have been considering whether a similar approach can be applied to design new superconductors with improved $T_{\rm c}$. It is well known that various bismuth (Bi) allotropes and compounds exhibit

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large λ comparable to those superhydrides, e.g., the Pb-Bi alloy ($T_c \sim 9 \text{ K}$, $\lambda \sim 2.3$) [11], amorphous Bi ($T_c \sim 6 \text{ K}$, $\lambda \sim 2.5$) [12], and the Bi-III phase $(T_c \sim 7 \text{ K}, \lambda \sim 2.75)$ [13]. The low-frequency phonons from heavy elements contribute to a large λ value but result in a low T_c due to much reduced ω_{\log} in comparison to superhydrides. Then, an interesting question arises, i.e., whether the incorporation of hydrogen in bismuth can raise T_c via enhancing ω_{\log} while maintaining the large λ . To this end, we recently initiated experimental explorations of bismuth hydrides in the Bi – H system under high pressures and obtained a novel molecular hydride superconductor, C2/c-BiH₄, with notably high $T_c = 91 \text{ K}$ at 170–190 GPa, setting a new benchmark for molecular hydride superconductors [14]. Unlike clathrate-type superhydrides that rely primarily on medium-to-high frequency hydrogen vibrations for EPC, the low-frequency translational vibrations of Bi atoms account for 38% of the total λ in the C2/c-BiH₄.

During the course of exploring the Bi – H systems, we observed another phase with a lower $T_c \sim 60$ K in addition to C2/c-BiH₄. Here we report on the discovery and comprehensive characterizations of this novel bismuth dihydride, Cmcm-BiH₂, synthesized at approximately 150 GPa and 2000 K. Interestingly, the structure of *Cmcm*-BiH₂ is featured by a three-dimensional bismuth open channel that encapsulates hydrogen molecules. Despite its low hydrogen content and the molecular nature of hydrogen in its structure, Cmcm-BiH₂ was found to exhibit an unexpectedly high T_c of 62 K at 163 GPa, marking the first instance of superconductor in MH₂-type metal dihydrides. Our calculations revealed that the electronic density of states (DOS) at the Fermi level, $N(E_{\rm F})$, is predominantly contributed by the Bi 6p orbitals and about 51% of λ arises from low-frequency vibrations of Bi atoms. This suggests that the observed high- T_c superconductivity in BiH2 is mainly driven by the covalent network of Bi atoms, while the high-frequency vibrations from hydrogen molecules that enhance ω_{\log} significantly boost the superconducting properties. These findings not only underscore the distinctive nature of Cmcm-BiH₂ compared to other high-pressure hydride superconductors but also offer valuable insights for the design and optimization of high- $T_{\rm c}$ hydride superconductors.

Based on our previous work on C2/c-BiH₄ [14], we explored the Bi – H system at lower pressures from 150 to 170 GPa aided by laser heating to ~2000 K and prepared a total of eight diamond anvil cells (DACs), using Bi foil and hydrogen sources, i.e., ammonia borane (BH₃NH₃), paraffin (C_nH_{2n+2} , n=18-39), and pure molecular hydrogen as reactants. The samples were labeled as s1 and s8 for structure analysis and s2 to s7 for transport measurements. The XRD data was also collected for s5 and s6 after resistance measurements as detailed in Table S1 [15]. Further details on sample preparation, structural and transport measurements, and DFT calculations are provided in Supplemental Material (SM) [15].

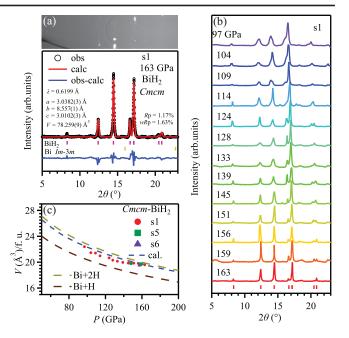


FIG. 1. XRD patterns and the result of Rietveld refinement for BiH₂ of s1 at 163 GPa in the *Cmcm* structure. Black circles: experimental data; red line: simulated XRD based on the structural model; purple and yellow vertical lines: Bragg diffraction positions of *Cmcm*-BiH₂ and *Im-3m*-Bi, respectively; blue line: the difference between the simulated and the original XRD. The upper panel shows the XRD image of corresponding powder XRD pattern with the incident x-ray wavelength of 0.6199 Å. (b) Representative integrated XRD patterns for BiH₂ of s1 upon decompression from 163 to 97 GPa. (c) Experimentally obtained volume per formula unit for BiH₂ plotted as a function of pressure. Red solid circles, green solid squares, and purple triangle represent experimental data for BiH₂. Theoretical EOS of BiH₂ is plotted as blue dashed line. Brown and yellow dashed line represents ideal mixtures of Bi + H and Bi + 2H.

Figures 1(a) and 1(b) present the synchrotron (x-ray diffraction) XRD patterns of s1 at selected pressures, where strong diffraction spots indicate a good crystallinity of the synthesized material. Comparison of the obtained XRD data with the Bragg positions of the cubic phase of bismuth clearly indicates that the bismuth metal almost fully reacted with the hydrogen released from the decomposition of BH₃NH₃. However, the XRD pattern cannot be described by any experimentally known or theoretically predicted bismuth hydrides [16]. Therefore, we employed the CALYPSO structure search method to identify the structure based on the obtained XRD pattern [17-19]. Detailed procedures of the structure search are provided in SM [15]. The results reveal that the diffraction peaks matched well with the BiH₂ phase with space group *Cmcm*. As shown in Fig. 1(a), Rietveld refinement using the Cmcm-BiH₂ structural model yields an excellent fit. The calculated equation of state (EOS) closely matches the experimental P-V data shown in Fig. 1(c), further confirming that the structure is *Cmcm*-BiH₂.

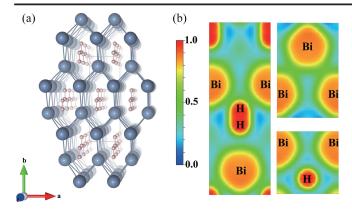


FIG. 2. (a) Crystal structure of Cmcm-BiH₂. Blue and pink spheres represent bismuth and hydrogen atoms, respectively. (b) The ELF projected on the plane (0 0 1) (left panel) for H–H contacts and the nearest Bi–H contacts, plane (0 2 1) for the nearest Bi–Bi contacts (up to the right panel), and plane (0 11 – 1) for the sub-closest Bi–H contacts (down to the right panel) of BiH₂.

Previous theoretical studies suggested that $P2_1/m$ -BiH₂ has the lowest enthalpy for this stoichiometry between 117 and 200 GPa [16]. However, compared with experimental data, the XRD pattern of $P2_1/m$ -BiH₂ shows significant discrepancies, as shown in Fig. S1 [15]. Then we carried out Gibbs free energy calculations under realistic high-pressure, high-temperature conditions to evaluate the relative stability of Cmcm-BiH₂ and $P2_1/m$ -BiH₂. The relative Gibbs free energy between Cmcm-BiH₂ and $P2_1/m$ -BiH₂ as a function of temperature and pressure are presented in the P-T phase diagram shown in Fig. S2 [15], which demonstrates that the formation of *Cmcm*-BiH₂ is thermodynamically favored at elevated temperatures within the pressure range of 150–200 GPa. Detailed procedures of the calculations are provided in SM [15]. As shown in Fig. 1(b), the XRD patterns upon decompression down to 97 GPa do not show any peak splitting or appearance of new diffractions. This confirms that Cmcm-BiH₂ synthesized at 163 GPa remains stable down to at least 97 GPa, below which the diamond anvils cracked during further depressurization process.

Figure 2(a) shows the crystal structure of Cmcm-BiH₂ for which the most prominent feature is the open-channel framework of Bi atoms filled by H₂-like molecules, forming a host-guest type structure like the high-pressure phase of Bi-III. The calculated structural parameters are provided in Table S3 [15]. It is well known that the Bi-III phase adopts an intriguing host-guest structure [37], characterized by a host lattice (I4/mcm) of covalently connected Bi atoms containing chains of guest Bi atoms (I4/mmm) with an incommensurate c axis relative to the host unit cell. It is noted that the Bi-Bi bond lengths in Cmcm-BiH₂ (2.8–2.9 Å at 150 GPa) are significantly shorter than those in the Bi-III phase (3.12–3.34 Å at 6.8 GPa) [37], which suggests the possible covalent bonding between Bi atoms in Cmcm-BiH₂. For the

 $\rm H_2$ -like molecular units, the H–H bond length of ~ 0.80 Å at 150 GPa is close to that in C2/c-BiH₄ [14], and slightly elongated compared to that of a hydrogen molecule (0.74 Å) at ambient pressure [38] but remains much shorter than those in monatomic hydrogen (0.98 Å at 500 GPa) [39]. This further elaborates the molecular hydride nature of the structure, with the Bi atoms exerting a precompression effect on the $\rm H_2$ molecules.

The bonding nature of Cmcm-BiH2 was first explored using electron localization function (ELF) and Bader charge analysis. At 150 GPa, the ELF value between the nearest-neighbor H-H atoms is approximately 0.97, confirming the presence of hydrogen molecules. In contrast, the ELF value between Bi-Bi atoms is relatively lower (~ 0.52) , indicating the weak covalent bonding, which facilitates the formation of metallic conductive channels between Bi atoms, Fig. 2(b). Bader charge analysis reveals an effective charge transfer of approximately 0.32e per Bi atom from Bi to H, which slightly elongated the H-H intramolecular bond length. To gain further insights into bonding behavior, we employed crystal orbital Hamilton population (COHP) analysis to quantify interatomic interactions. The integrated COHP (ICOHP) and negative projected COHP (-pCOHP) data for intramolecular H-H, Bi-Bi, and Bi-H interactions at different pressures are presented in Table S4 and Fig. S3 [15]. The strong intramolecular H-H interaction observed through COHP analysis confirms the molecular hydride character of H₂ molecules. Additionally, the H-H bonds exhibit an antibonding character, as indicated by the negative –pCOHP at the Fermi level, supporting that electron transfer from Bi atoms to the H₂-like molecular units fills the antibonding orbitals. The ICOHP analysis reveals weak covalent interactions between Bi atoms, forming a distorted hexahedral prism with Bi atoms alternately arranged in triangular prismatic six-coordination units. Such configuration results in the formation of a three-dimensional bismuth openchannel structure. Unlike the clathrate hydride superconductors where a weak covalent hydrogen clathrate serves as the host and metal atoms act as guests, Cmcm-BiH₂ exhibits a three-dimensional Bi atom framework with weak covalent Bi-Bi interactions as the host, encapsulating H₂-like molecules as guests. Such a bismuth host framework closely resembles the Bi-III phase, which was recently revealed to feature a significant large EPC, $\lambda \sim 2.75$ and non-Fermi-liquid behaviors prior to the superconducting transition at $T_c \sim 7$ K [13]. In this regard, it is essential to conduct resistance measurements to examine the superconducting properties of *Cmcm*-BiH₂.

Following the synthesis conditions for s1, we compressed s2 to 155 GPa at room temperature and subsequently laser heated it to ~2000 K, after which the pressure decreased to 152 GPa. We conducted independent experiments on s3, s4, and s5 under similar experimental conditions to check the reproducibility. The results are

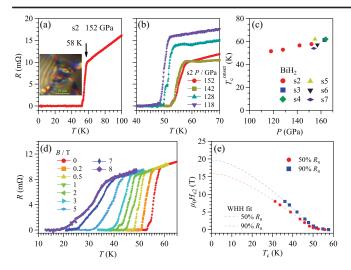


FIG. 3. (a) Temperature-dependent resistance of s2 at 152 GPa. The left inset shows the configuration of sample in chamber and the permutation of electrodes. (b) Temperature-dependent resistance of s2 under selected pressures. (c) Pressure dependent $T_{\rm c}$ of Cmcm-BiH₂ in different runs. (d) Temperature dependences of resistance under various magnetic fields for s2 at 152 GPa. (e) Temperature dependence of $\mu_0 H_{\rm c2}({\rm T})$ fitted by the WHH two-band model.

depicted in Figs. 3(a) and S4 [15], respectively. Figure 3(a) presents the temperature-dependent resistance R(T) of s2 at 152 GPa, with the inset showing a photograph of the sample in the DAC used for resistance measurements. The R(T) data clearly demonstrate a superconducting transition, as evidenced by a sharp drop at 58 K, reaching zero resistance at around 53 K. For s2, we measured R(T) under different pressures ranging from 152 to 118 GPa, Fig. 3(b). The $T_{\rm c}$ decreases upon decompression, reaching about 51 K at 118 GPa for s2. Upon further decompression, the diamond anvils were cracked. The pressure dependence of $T_{\rm c}$ obtained from resistance measurements on these samples is depicted in Fig. 3(c). The highest $T_{\rm c}$ observed in this experiment is 62 K, recorded in s3, s4, and s5 at 163 GPa, 164 GPa, and 155 GPa, respectively.

To further substantiate the occurrence of the superconducting transition, we examined the sample's response to external magnetic fields. Figure 3(d) shows the R(T) of s2 under various magnetic fields. As can be seen, the superconducting transition is progressively suppressed with increasing magnetic fields, in line with the expected downward shift behavior of a superconductor. The values of T_c at 50% R_n and 90% R_n were utilized to establish the relationship between $\mu_0 H_{\rm c2}(T)$ and temperature. As shown in Fig. 3(e), within the magnetic field range of 0-8 T, $\mu_0 H_{c2}(T)$ exhibits a distinct nonlinear behavior, characterized by a notable upward curvature, a feature commonly observed in two-band superconductors. By fitting the experimental $\mu_0 H_{c2}(T)$ data using the Werthamer-Helfand-Hohenberg (WHH) two-band model in the dirty limit [20], we determined the zero-temperature

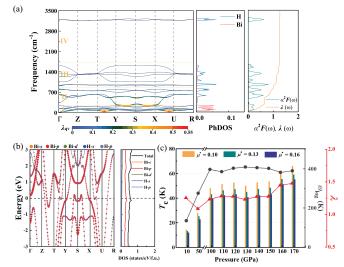


FIG. 4. (a) Calculated phonon dispersion, projected PhDOS, Eliashberg phonon spectral function $\alpha^2 F(\omega)$, and integrated electron-phonon coupling $\lambda(\omega)$ for *Cmcm*-BiH₂ at 150 GPa. (b) The calculated electronic band structure and projected DOS of *Cmcm*-BiH₂ at 150 GPa. (c) The calculated electron-phonon coupling constant λ , logarithmic average frequency $\omega_{\rm log}$, and $T_{\rm c}$ for $\mu^*=0.1,~0.13,~$ and 0.16 at the pressure range of 10 to 170 GPa.

 $\mu_0 H_{\rm c2}(0)$ to be 15.8 T for $T_{\rm c}^{50\% Rn}$ and 19.7 T for $T_{\rm c}^{90\% Rn}$, respectively (Table S6 [15]). These values are much smaller than the weak-coupling Pauli limit of $\mu_0 H_{\rm p}^{\rm BCS} = 1.84 T_{\rm c} \approx 102.7$ and 106.2 T. Details of the fitting process are provided in SM [15]. Additionally, after detecting superconductivity in sample s5, we conducted *in situ* XRD measurements to verify the phase identity. A comparison of the XRD pattern of s5 at 160 GPa with that of s1 at 163 GPa, Fig. S5 [15], further confirmed that the superconducting sample is indeed the *Cmcm*-BiH₂.

The observed T_c of 62 K at 163 GPa for *Cmcm*-BiH₂ marks the first instance of a superconductor among the MH₂type metal dihydrides. To gain deeper insight into the superconducting properties of Cmcm-BiH₂, we further analyzed its phonon density of states (PhDOS), Eliashberg phonon spectral function $\alpha^2 F(\omega)$, and EPC constant $\lambda(\omega)$. As shown in Fig. 4(a), the absence of imaginary frequencies in the phonon spectra of Cmcm-BiH₂ at 150 GPa indicates its dynamical stability. The calculated PhDOS can be divided into two regions: low-frequency vibrations from Bi atoms and mid- to high-frequency vibrations from hydrogen atoms. The calculated λ and ω_{\log} for $\textit{Cmcm-BiH}_2$ is around 1.27 and 37.19 meV at 150 GPa. The calculated T_c of Cmcm-BiH₂ by using the Migdal-Eliashberg equations with $\mu^* = 0.1$ is about 54 K at 150 GPa, which is close to our experimental observation.

As illustrated in Fig. 4(a), the Bi-derived and H-derived phonon modes are well separated. Thus, the total EPC constant λ can be decomposed into atom-specific

contributions, $\lambda_{\rm Bi}$ and $\lambda_{\rm H}$. Specifically, the phonon dispersion curves are divided into four regions according to the vibrational frequencies, i.e., I, II, III, and IV in Fig. 4(a), and the detailed vibrational modes of each region are demonstrated in Fig. S6 of SM [15]. The region I in the vibrational frequencies from 0 to 250 cm⁻¹ is primarily attributed to Bi atoms, yielding an integrated $\lambda_{\rm Bi} \sim 0.65$, which contributes approximately 51% of the total λ in Cmcm-BiH₂. In this region, the phonon modes are predominantly associated with the movements of Bi atoms, including in-phase vibrations at 0 cm⁻¹ (acoustic branch) and bending vibrations within the Bi lattice at $122 - 250 \text{ cm}^{-1}$ (relative vibrations between pairs of Bi atoms). Regions II and III in the frequencies ranging from 250 to 1640 cm⁻¹ contribute approximately 46% to EPC. These regions involve the vibrations of H ions perpendicular to Bi-H bonds and the in-phase vibrations of H₂ molecules, respectively. In region IV, the vibrations of the hydrogen atoms along the H-H bond direction contribute only 3% to EPC. As shown in Fig. 4(b), the calculated electronic structure exhibits metallic behavior with a relatively high $N(E_{\rm F})$. The phonon dispersion demonstrates that the strongest EPC contributions arise along the T-Y and X-Upaths, with additional significant contributions from the Y-S-X path. Notably, the electronic bands near the Fermi level along these paths are predominantly derived from p orbitals of Bi. Consistently, the projected DOS analysis shows that p states of Bi account for the majority (\sim 79%) of $N(E_{\rm F})$, ensuring strong coupling with low-frequency phonon modes involving Bi vibrations. However, the contribution from hydrogen-derived states is relatively small. This is consistent with the behavior typically observed in molecular hydrides, where the electronic states near the Fermi level are dominated by the heavy element sublattice.

According to the McMillan-Hopfield theory [21,22], the large $\lambda_{\rm Bi}$ mainly originates from the high $N(E_{\rm F})$ (Bi) contributions, despite the smaller I^2 and larger $M\omega_2^2$ values compared to H (Table S5 [15]). In contrast, in covalent and clathrate-type hydride systems, both the $N(E_{\rm F})$ and the EPC strength are typically dominated by hydrogen-derived electronic and vibrational contributions [9,10,40]. Although hydrogen makes a relatively small contribution to $N(E_{\rm F})$ in Cmcm-BiH₂, the presence of H₂ molecules plays a crucial role in boosting ω_{\log} while maintaining the strong λ , thereby further enhancing superconductivity of BiH₂. This essential role of incorporating of H_2 molecules explains why the T_c of Cmcm-BiH₂ (\sim 62 K) is significantly higher compared to the Bi-III phase (\sim 7 K) [13], even though the total λ of *Cmcm*- BiH_2 is lower than that of Bi-III (2.75). Notably, the ω_{log} of Cmcm-BiH₂ is around ten times higher than that of Bi-III (3.73 meV). Such synergy leads to an increase in ω_{log} , ultimately resulting in a substantial elevation of T_c . Remarkably, Cmcm-BiH₂ remains dynamically stable down to 10 GPa without imaginary frequencies on the phonon spectrum (Fig. S7 [15]), a significantly lower pressure than typically required to stabilize high-pressure hydride superconductors. As pressure decreases, $T_{\rm c}$ slowly and monotonically decreases, remaining around 48 K at 100 GPa. However, below 100 GPa, the $T_{\rm c}$ decreases rapidly, reaching approximately 17 K at 10 GPa [Fig. 4(c)].

It should be noted that the decomposition of BH₃NH₃ is a complex multistep process and its direct contact with Bi during high-pressure/high-temperature (HPHT) synthesis might potentially produce other phases containing B and/ or N. To address this issue, we conducted comprehensive control experiments as follows. First, we test directly the possible reaction between Bi and BN by treating their mixtures under comparable HPHT conditions as those for the BiH₂ synthesis. Synchrotron XRD measurements (Fig. S8 [15]) revealed no formation of borides, nitrides, or related phases, confirming that Bi and BN do not react under these conditions. Second, we carried out HPHT syntheses by using alternative hydrogen sources free of boron and nitrogen, i.e., paraffin and pure molecular hydrogen. Synchrotron XRD confirmed that the same crystalline phase was consistently obtained using these alternative hydrogen sources, identical to that synthesized with BH₃NH₃ (Figs. S8 and S9 [15]). Furthermore, electrical transport measurements on two paraffin-derived samples revealed superconducting transitions at 57 K and 54 K, respectively (Fig. S8 [15]), in excellent agreement with the pressure-dependent T_c values observed in BH₃NH₃-derived samples [Fig. 3(c)]. Taken together, these two sets of experiments conclusively rule out any chemical incorporation or reaction involving B or N with the superconducting phase, providing unambiguous confirmation that the synthe sized superconducting compound is indeed *Cmcm*-BiH₂.

According to recent theoretical study [41], the $T_{\rm c}$ of BiH₂ will not exceed 58 K, due to the relatively high atomic mass ratio between Bi and H. Thus, the experimental observation of a high T_c up to 62 K in Cmcm-BiH₂ is quite unexpected for a heavy metal dihydride with such a low hydrogen content. It is well known that MH₂-type dihydride is one of the most common forms of metal hydrides at ambient pressure, typically characterized as insulators or poor metals. In these conventional MH2 compounds, the structure is often ionic, with hydrogen atoms occupying interstitial octahedral or tetrahedral sites within densely packed metal lattices [42]. This configuration leads to the localization of hydrogen electrons at low energy levels, contributing minimally to $N(E_{\rm F})$. To our knowledge, metal dihydrides MH2 have never been found to exhibit superconductivity. Unlike previously reported high-T_c hydride superconductors [9,10,40], Cmcm-BiH₂ represents a novel structural type where a three-dimensional bismuth host framework with weak covalent Bi-Bi interactions encapsulates H₂-like molecules as guests. DFT calculations further reveal that the unique bismuth open-channel framework forms metallic conduction channels, dominating the electronic states near the Fermi level and playing a crucial role in its superconducting properties. These findings underscore the pivotal role of nonhydrogen elements in hydride superconductors and offer fresh insights and guidance for developing high- $T_{\rm c}$ hydride superconductors at lower pressures.

In summary, we have successfully synthesized a bismuth dihydride superconductor, Cmcm-BiH₂, featuring a unique host-guest structure. In Cmcm-BiH₂, weak Bi-Bi covalent bonds form a three-dimensional bismuth open-channel framework that encapsulates H₂-like molecules as guests. Cmcm-BiH₂ exhibits superconductivity with T_c of 62 K at 163 GPa, establishing the first reported instance of superconductor among MH₂-type metal dihydrides. Detailed analyses reveal that the bismuth open-channel framework plays a dominant role in its superconducting properties. Our Letter not only uncovers a novel hydride structure, broadening the scope of experimentally realized metal hydrides, but also provides valuable insights for the design and optimization of new hydride superconductors.

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Data availability—The data that support the findings of this article are not publicly available. The data are available from the authors upon reasonable request.

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